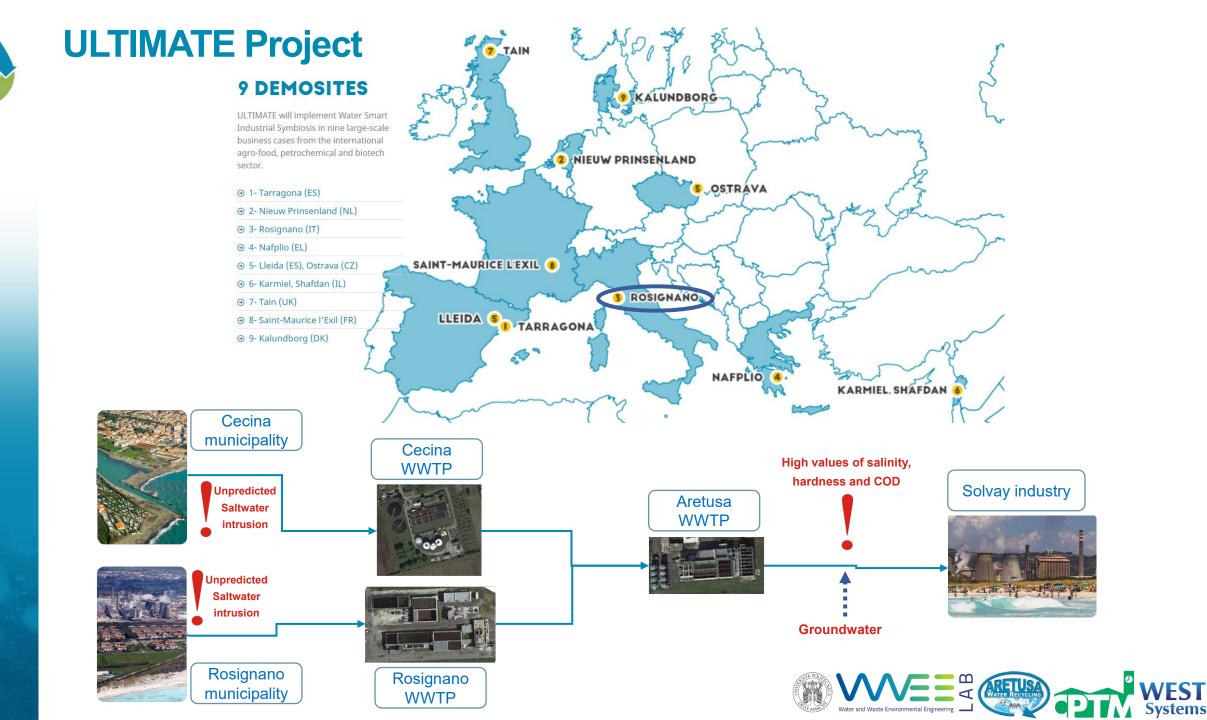


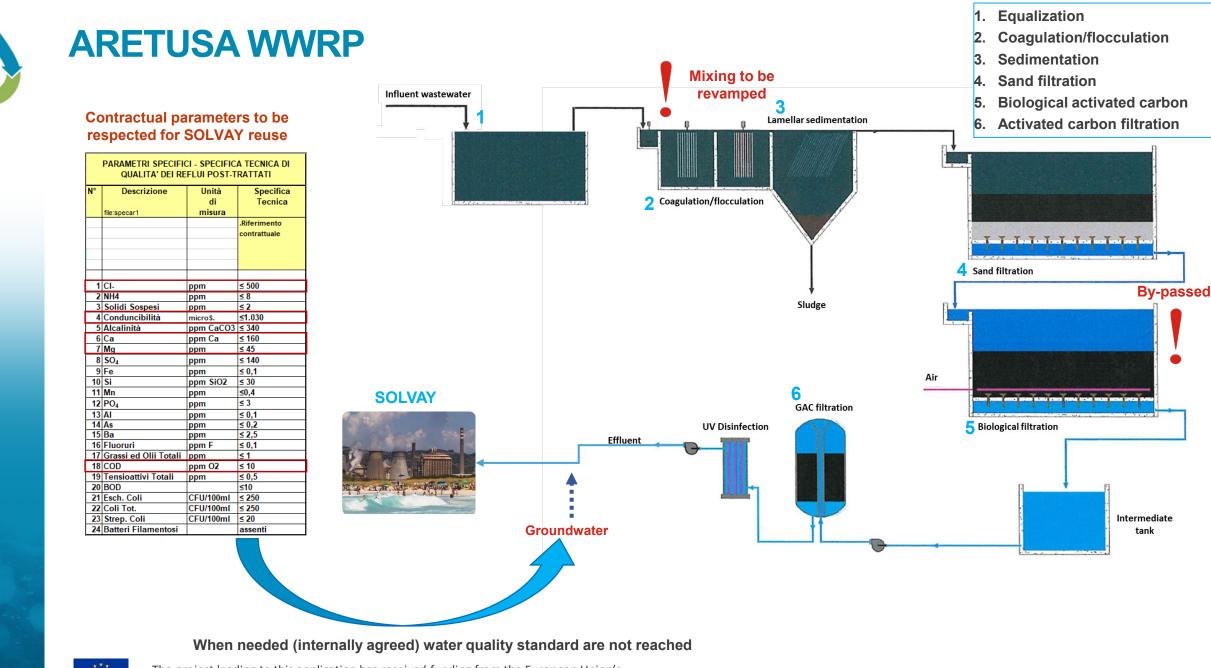
WATER SMART INDUSTRIAL SYMBIOSIS

Use of by-products of local industries for wastewater treatment in Rosignano

F. Fatone, F. Rossi

June, 23rd 2021





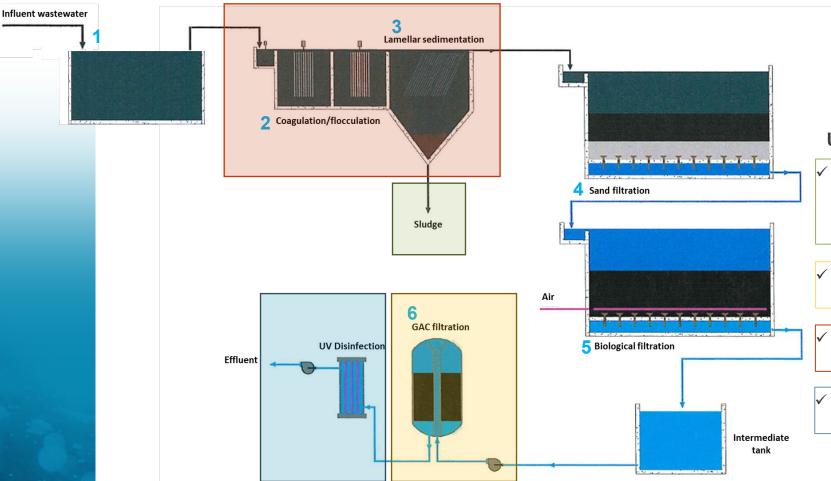
vstems

The provide the providet the pro

The project leading to this application has received funding from the European Union's Horizon 2020 research and innovation programme under grant agreement No 869318



ULTIMATE Solutions



- 1. Equalization
- 2. Coagulation/flocculation
- 3. Sedimentation
- 4. Sand filtration
- 5. Biological activated carbon
- 6. Activated carbon filtration

ULTIMATE SOLUTIONS

The possibility to re-use the chemical (alum/ferric) sludge from coagulation/flocculation in the WWRP will be analysed and potential users will (potentially) be identified via the Alu Circles initiative.

 A pilot scale adsorption system will be tested with alternative sludge-biowaste-originated GAC.

✓ Mineral by-products will be used as alternative coagulants and/or adsorbent.

Possible residual disinfectant agents will be tested in order to improve disinfection.





Possibility to re-use the chemical (alum/ferric) sludge and potential users identified via the Alu Circles initiative

The ALU Circles initiative (https://www.alliedwaters.com/project/alucircles/) is addressing the challenge to convert the alum sludge from one-off use of material into a sustainable solution, such as upcycling or recycling, at a lower cost.

PLANNED ACTIVITIES

- 1. Mass flow analysis : Production of Al/Fe sludge in Italy and contacts with major water utilities (problem owners)
- 2. Fate of the Al/Fe sludge: Current disposal/reuse routes and related costs
- 3. Characterization of AI/Fe sludge: Physical-chemical parameters
- 4. Preliminary assessment of possible applications: absorbents/coagulants in wastewater treatmnent, geotechnical, geoenvironmental and building sectors
- 5. Experimental tests: Mechanical properties in construction materials / removal efficiencies in tertiary wastewater treatment
- 6. Local and regional impact by LCA and LCC of the most promising solutions



5



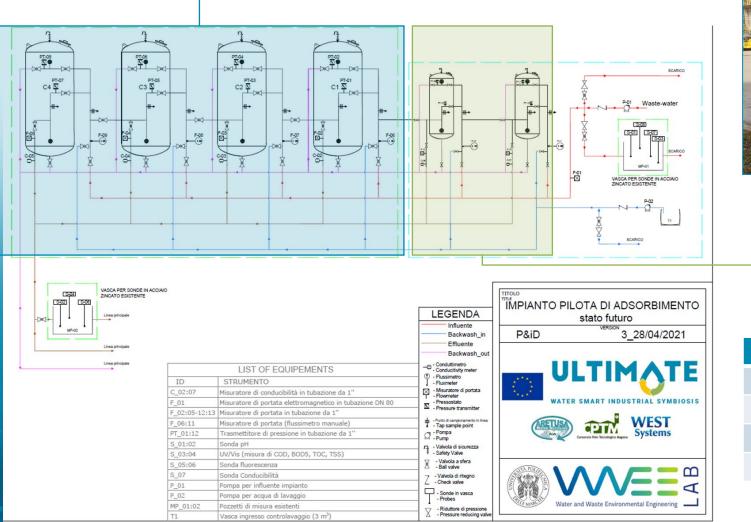
THE UNIVERSITY

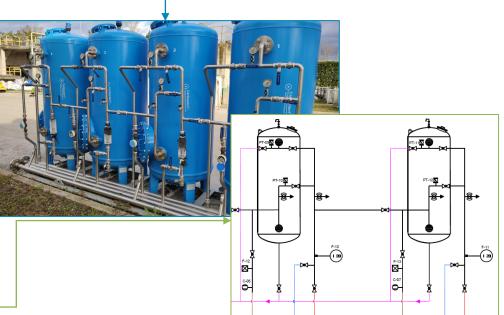
aqua

()F

AUSTRALIA

A pilot scale adsorption system will be tested with alternative GAC.





	Bigger columns	Smaller columns
Numbers	4	2
Volume (m3)	0.6	0.11
Diameter (m)	0.8	0.32
Material (kg)	325	60





\bigcirc

A pilot scale adsorption system will be tested with alternative GAC.

ULTIMATE INNOVATIONS

 \checkmark Monitoring system \rightarrow sensors and probes

Influent and effluent from the pilot will be monitored with:

- Conductivity
- pH
- UV/Vis (COD, BOD5 and TOC)
- Fluorescence





ID	INSTRUMENTS	Q. ty	Position	Characteristics
PT-01:08	Pressure transmitter	12	Influent and effluent from each column	0 – 1 bar
F-02:05	Electromagnetic water meter	6	Effluent from each column	0 – 10 m3/h
C-02:05	Insertion inductive conductivity digital probe	6	Effluent from each column	250 μS - 2,5 S/cm
S-01:02	pH meter	2	Influent and effluent	pH:0-14; T=-5° C - 50° C
S-03:04	UV/Vis meter (254 nm)	2	Influent and effluent	Measuring cell: 5 mm Measuring range: 0.1- 600.0 m-1 Calibrable on the COD and TOC parameters Compensation: 550 nm Cleaning system: automatic by wiper Measuring interval:> 1 min
S-05:06	Fluorimeter	2	Influent and effluent	filter band for wavelengths centered on ex/em 345/440 nm
S-07	Conductivity meter	1	Influent	250 μS - 2,5 S/cm
F-01	Electromagnetic water meter	1	Influent	0 – 50 m3/h



✓ Innovative material as adsorbents → EXPERIMENTAL ANALYSIS ONGOING!







Mineral by-products will be used as alternative coagulants and/or adsorbent



In the past 140 t per year of aluminium polychloride, and 12 t per year of polyelectrolyte were used. At the moment no reagent is used, and some mixers needs revamping



ULTIMATE will test and assess the possibility of using industrial by-products EXPERIMENTAL ANALYSIS ONGOING

COAGULATION/FLOC CULATION UNIT	Rapid mixing (coagulation)	Medium agitation (primary flocculation)	Slow mixing (secondary flocculation)
Number of tanks	2	2	2
Volume (m3)	5.73	73	72





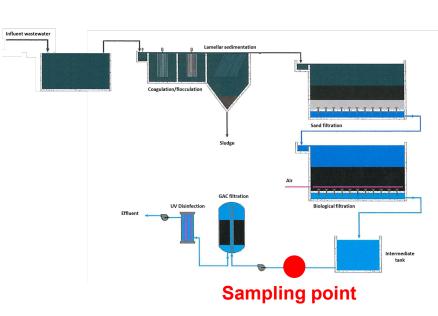


Characterization of the water sample collected in May 2021

Sample was collected at Aretusa plant before the final GAC stage in order to find a higher COD concentration.

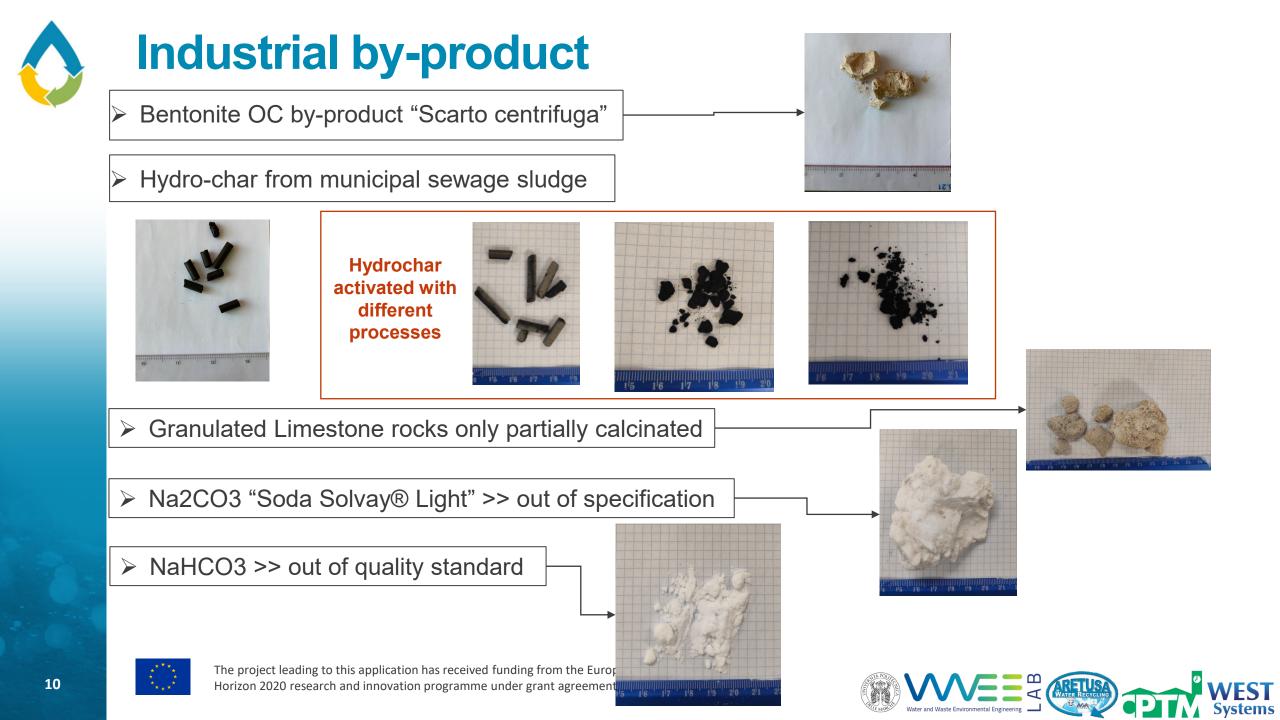
Measured values are compared with the average, minimum and maximum from historical data, and with the Aretusa specification.

		Average	Min	Max	Measured	Aretusa-Solvay contractual constraint
рН	-	7.6	7.0	7.9	7.7	
Electrical conductivity	μS/cm	2580	1010	7340	2134	1030
CI-	mg/l	568	159	2079	398	400
NO3	mg NO3/I	57	8	128	45	
PO4	mg PO4/I	/	5	15	10	
SO4	mg/l	/	100	150	132	
Са	mg/l	/	140	180	145	
Mg	mg/l	/	45	60	54	
Κ	mg/l	/	13	18	17	
Na	mg/l	/	225	305	232	
COD	mg O2/I	25	10	88	20	10









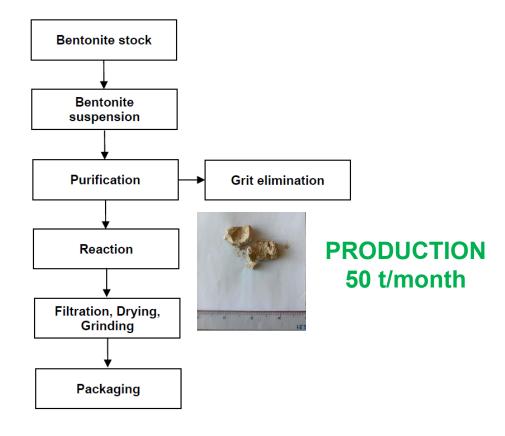


Bentonite OC by-product "Scarto centrifuga" – LAV1

This by-product comes from Laviosa Chimica Mineraria SpA plant (located about 30 km from Aretusa).

Laviosa activities consists in extracting, processing and distributing industrial mineral products, in particular bentonitic products and special 'modified' bentonitic products called "Organo-clay".

From the necessary purification stages in the organo-clay production process comes this 'grit' that is poor in bentonite but rich in zeolite and other silicates: it comes from a decanter separation and looks like a semi solid sludge.



Technical specifications

Le analisi effettuate di diffrazione ai raggi X dimostrano che la fase più abbondante risulta quella zeolitica e le altre fasi sono silicate quali calcite, plagioclasio, miche ed altri riportati nella tabella sottostante:

	Fase	Quantità [%]
	Zeoliti	33
	Calcite	28
KRD	Plagioclasio	15
	Quarzo	10
	Mica	8
	K-feldspato	6

L'umidità risulta mediamente del 35 % in massa e la perdita alla calcinazione del 11,67 %.

La composizione degli ossidi invece risulta come nella seguente tabella:

	B	
	Analita	Quantità [%]
	Na ₂ O	1,80
	MgO	0,86
	Al ₂ O ₃	11,75
Oxides	SiO ₂	59,05
	P2Os	0,35
composition	K ₂ O	2,76
	CaO	9,85
	TiO ₂	0.30
	MnO	0.40
	Fe ₂ O ₃	1,22





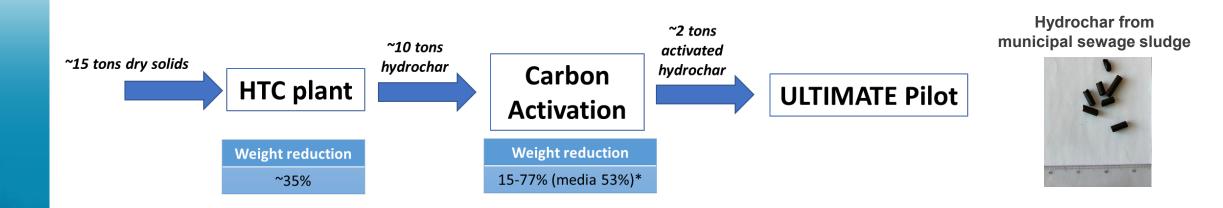
\diamond

Adsorbent materials for COD removal

Activated Hydro-char from municipal sewage sludge

Hydrochar (HC) was supplied by the **Spanish company Ingelia SL**: this biocarbon is the product of a **Hydrothermal Carbonization (HTC)** process which, operating at relatively low temperature and pressure conditions in the presence of liquid water, **converts the biomass or municipal sewage sludge, in a carbonaceous solid (bio-lignite).**

Operating on small quantities, in CPTM laboratory, the **HC was activated** in order to be tested as an adsorbent material. **The activation procedures**, suggested by the scientific literature, **were physical (thermal) and chemical**.









Adsorbent materials for COD removal Activated Hydro-char from municipal sewage sludge

Physical activation – ATT1

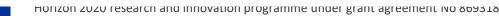
- Heating of the char pellets in a tubular oven up to 700°C (5°C/min) with N_2 purging.
- CO_2 flushing and isotherm for 2 hr.
- Cooling of the tubular furnace in N₂ purging.

Chemical activation – ATT4

- Impregnation of char pellets in <u>KOH aq. solution</u> (KOH to char ratio: 1:1) at 60°C for 6 hr.
- Drying of the impregnated char at 105°C.
- Heating in a tubular oven up to 600°C (5°C/min), isotherm at 600°C for 1 hr and cooling (5°C/min) with N₂ purging.
- Washing with 1M HCl and demi water (up to pH 7).
- Drying at 105°C until constant weight.

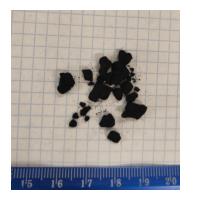
Chemical activation – ATT5

- Mixing of the char pellets (previously grounded) with <u>KOH in flakes</u> (KOH to char ratio: 1:1).
- Heating in a tubular oven up to 600°C (5°C/min), isotherm at 600°C for 1 hr and cooling (5°C/min) with N₂ purging.
- Washing with 5M HCl and demi water (up to pH 7).
- Drying at 105°C until constant weight.

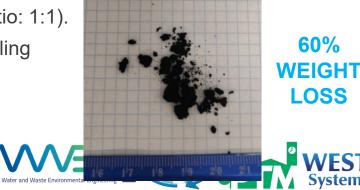




47% WEIGHT LOSS



55% WEIGHT LOSS



Commercial activated carbon – CA1



Review : 6 Date of issue : February, the 1# 2018

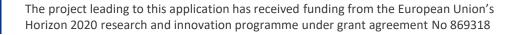
Technical Data Sheet

FILTERCARB GCC 8X30

Granular Activated Carbon obtained from specially selected grades of coconut shell, specially designed to remove pollutants from water, odours and bed tests from beverages like edible oils, wine, beer, ex...

SIANDAI	STANDARD STECIFICATIONS					
Specification	Units	Values	Methods			
BET Surface	m²/g	1100-1150	BET N2			
Ash content	%	2-4	CEFIC 1986			
Bulk content	g/cm ³	0,45-0,50	CEFIC 1986			
Moisture	%	5	CEFIC 1986			
Iodine index	mg/g	1000-1100	CEFIC 1986			
CTC	%	55-65	Method on request			
Hardness	%	95	Method on request			
Chlorine half-length value	cm	3	Method on request			
Grain size distribution			Sieving analysis			
> 8 mesh (2,36 mm)	%	5				
< 30 mesh (0,60 mm)	%	5				
Uniformity coefficient		max 1,7				

STANDARD SPECIFICATIONS







Adsorbent materials characterization

Leaching test

Test followed the EN 12457-2:2002 for the waste characterization (liquid to solid ratio of 10 L/kg).

Concentrations are expressed as ppm in the leached solution.

- NEED OF PRE-TREATMENT (WASHING) OF THE HC
- FURTHER TEST FOR ADSORPTION IN THE INTEGRATED SYSTEM (ROSIGNANO-CECINA WWTP)

	LAV1	НС	HC ATT1	HC ATT4	HC ATT5
F1	< 0.1	54.2	0.2	0.4	< 0,1
CI-	2,5	44.4	62.5	26.4	1.7
NO3 ⁻	0.2	1.3	< 0,1	< 0,1	< 0,1
PO4	3.2	38.5	< 0,1	54.3	8.4
SO4	32.9	147.3	143.9	110.2	103.4
COD	81	4200	< 15	< 15	< 15

	LAV1	HC ATT1	HC ATT4	HC ATT5	CA1
Specific surface area (m²/g)	6	117	449	752	1100÷1150
Specific pore volume (cm ³ /g)	0.003	0.055	0.214	0.359	/
Average pore radius(Å)	50.23	13.61	15.16	16.08	/



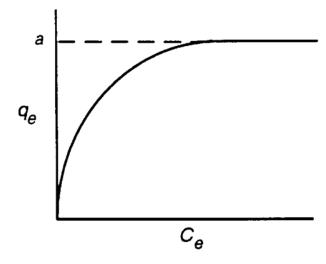
BET analysis



Preliminary adsorption tests with Methylene Blue (MB) solution

MB solutions are commonly used for comparative adsorption tests on activated carbon and they simulate chemical substances with an effective molecular diameter of 15 Å.

The Adsorption capacity (q_e) is the amount of adsorbate taken up by the adsorbent per unit mass of the adsorbent at equilibrium conditions. It changes with the final equilibrium concentration of the adsorbate in the solution (c_e) and an 'isotherm curve' can be realized for each material tested.



This is a 'Langmuir isotherm' that describes the most common adsorption mechanism of an activated carbon.





Preliminary adsorption tests with Methylene Blue (MB) solution

CPTM is currently completing these preliminary tests in order to compare the different adsorbent materials.

PRELIMINARY RESULTS >> TEST STILL ONGOING!

	С _е (mg _{мв} /L)	Q_e (g/kg)
LAV1	0.6	11
ATT4	0.1 4.0	305 321
CA1	0.1 0.9	304 383

LAV1 material (Bentonite OC by-product by Laviosa) **shows a lower adsorption capacity** compared to the chemically activated HC ATT4 and to the commercial activated carbon.

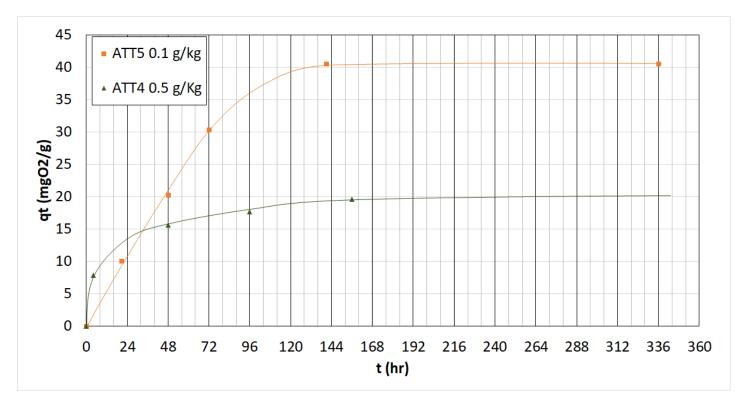




Adsorption tests with Aretusa water sample (COD reduction) ARETUSA sample \rightarrow COD = $\sim 20 \text{ mgO}_2/\text{L}$.

Kinetic adsorption tests:

Here below an example of two kinetic adsorption curves: the scope of these tests is to evaluate the minimum time necessary to reach the equilibrium state and find the Adsorption capacity (qe) of the material.







Adsorption tests with Aretusa water sample (COD reduction)

CPTM is currently completing these preliminary tests in order to compare the different adsorbent materials that can be used for the COD reduction in the Aretusa water.

PRELIMINARY RESULTS >> TEST STILL ONGOING!

	C _e (mgO₂)/L	₽ _e (g/kg)
LAV1	16	4.0
ATT4	7 17	19.6 20.0
ATT5	9 15	40.5 40.0

These tests confirmed that **LAV1 material** (Bentonite OC by-product by Laviosa) **shows a lower adsorption capacity compared to the chemically activated HC** ATT4 and ATT5.

ATT5 shows a better adsorption capacity than ATT4 probably caused by the higher specific surface area previously described (752 and 449 m2/g).





Preliminary conclusions

- ✓ Both from the preliminary tests conducted with MB solutions and from those conducted with the real water sample, it can be anticipated that the adsorption capacity of LAV1 is much lower than that of activated HCs and unfortunately does not probably justify its possible use as an adsorbent material.
- ✓ The different activated HCs will be compared with the commercial activated carbon.







Softening tests

HARDNESS (mg _{CaCO3} /L)				
Direct measure*	Aretusa-Solvay contractual limit			
520	340			
*Direct measure with EDTA complexometric titration				

Hardness overcome the contractual limit requested by SOLVAY

Specific analyses have shown that about 50% of the total hardness is temporary hardness essentially due to Ca(HCO3)2 while the remaining part of 50% is due to the permanent hardness which is mainly composed of chloride, sulphate and nitrate of Mg.

SOFTENING TECNIQUES: lime and the lime-soda ash process.

- ✓ "Precotto": granulated limestone rocks only partially calcinated and slacked, with a declared content of $Ca(OH)_2$ of about 9%.
- ✓ Na₂CO₃ "Soda Solvay® Light" product that resulted to be out of specification.









Softening tests

Solvay by-product and OOS were compared to pure laboratory lime and soda ash.

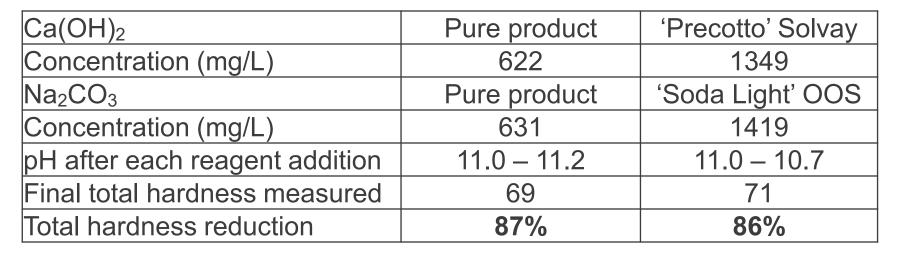
Lime process



Ca(OH) ₂	Pure product	'Precotto' Solvay
Concentration (mg/L)	583	1074
pH after the reagent addition	10.8	10.5
Final total hardness measured	277	238
Total hardness reduction	46%	54%

Lime - Soda ash process











- ✓ Solvay annual production of 'Precotto' and of out-of-spec 'Soda Light' is absolutely sufficient to treat the total Aretusa water output.
- ✓ The preliminary tests conducted shows a very good total hardness reduction.
- ✓ Specific tests will be carried out to optimize the exact dosage of the two reagents and to evaluate the generated sludge amount.



